# **RHODUNA® 271**

## **Operating Instructions**

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Electrolyte for selective decorative pencil rhodium-plating (brush rhodium-plating)

- abrasion-resistant, light and bright coatings
- operating temperature 20 °C to 40 °C
- deposition speed up to 0.2 μm/min
- ready for use with 2 g Rh/100 ml
- · high covering power

## **Electrolyte Characteristics**

Pencil plating is a special type of brush plating. With pencil rhodium-plating, small area elements of parts are selectively rhodium-plated by means of direct current through a wiping touch with a tampon tip soaked with the respective rhodium electrolyte.

The process mainly serves for applying decorative patterns on e.g. jewellery, spectacles, tableware, and things like that.

Although for decorative purposes mostly only layer thicknesses of 0.1 - 0.2  $\mu$ m are applied, the process also supplies bright, pure-white coatings up to approx. 1  $\mu$ m. RHODUNA® 271 is distinguished by its high covering power and high plating speed.

Electrolyte

pencil rhodium-plating electrolyte

type:

Rhodium 2 g Rh/100 ml

content:

pH-value: < 1, strongly acidic

Temperature: room temperature (20 - 40 °C) Voltage: 8 - 10 volts (max. 12 volts)

Deposition

speed:

up to 0.2 μm/min

#### **Coating Characteristics**

Coating: rhodium
Fineness: 99.9 % Rh
Colour: white

Hardness: approx. 800 HV

## **Basic Equipment**

The basic equipment required for pencil rhodiumplating consists of:

a) A tampon pencil including a supply of tampon tips.

This pencil is normally a plastic holder with a connection to the rectifier. The absorbent and exchangeable tampon tips are inserted into a suitable contact socket at the front end of the holder.

- b) Pencil rhodium-plating electrolyte RHODUNA® 271.
- c) Rectifier with adjustable voltage up to at least 10 volts.

We recommend our rectifiers

TG 511 (10 V/10 A) TG 2011 (10 V/20 A) or TG 5050 (20 V/50 A).

## Form of Supply

 a) RHODUNA<sup>®</sup> 271, ready for use (2 g Rh/100 ml) Solution ready for use containing 2 g of rhodium per 100 ml. Delivered in 100 ml units. Storage stability: min. 2 years Store in a cool (10 – 30 °C) place!

Special replenishments and corrections are not necessary because the solution is used up during application.

### **Operating Conditions**

Rhodium 2 g Rh/100 ml

content:

Operating room temperature (20 - 40 °C)

temperature:

pH-value: < 1, strongly acidic

Voltage: 8 - 10 volts (maximally 12 volts)

Deposition Approx. 0.1 µm can be reached if

speed: 1 cm² is treated for about 30 seconds

while the movement is constant.

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In addition the voltage can be increased to maximally 12 volts and the electrolyte temperature to 40 °C. Higher voltages must not be applied because then the tampon tips will be rendered unusable due to melting.

In any case take care that the tampon tip is always well soaked with the electrolyte. Usually it is sufficient to dip the pencil into the electrolyte approx. 2 - 3 times per minute.

The specified speed may only serve as a rough approximate value because the deposition speed is influenced by the individual carrying out of the work (movement, pressure, dipping frequency).

## **Special Process Hints**

Procedure:

The cable fixed to the tampon pencil is connected to the positive pole of the rectifier. The negative pole is connected to the workpiece to be treated by means of a second cable and e. g. a crocodile clip.

Suitable are rectifiers supplying at least 8 - 10 volts. The voltage should be adjustable and measurable, as e.g. with rectifier TG 511 (up to 10 V).

The tampon tip should always be well soaked with the electrolyte. If the tip is dry, remove it from the holder and dip both ends into the electrolyte, one after the other, until it is completely saturated. Then the tip is again deeply inserted into the contact socket of the holder.

<u>Caution:</u> The electrolyte is strongly acidic! Handle the tips with tweezers and rubber gloves! For different types of electrolytes different tips must be used by all means!

Pretreatment:

As with any other plating treatment, the parts first have to be cleaned thoroughly. This is done in the usual way by soak cleaning or ultrasonic pre-degreasing and electrolytic degreasing. After intermediate rinsing, briefly dip in 5 % sulphuric acid and rinse again. Afterwards, rinse in deionized water.

If fine structures with borders as exact as possible are to be applied, the surface of the part should be dried rapidly or the adhering water film should at least be spun off. If the surface of the part is wet, the electrolyte film will easily run during the treatment and thus lead to blurred borders.

Pencil rhodiumplating: The connections to the part (cathode, negative pole) and to the pencil (anode, positive pole) have been made. Turn on the current and adjust the voltage. After dipping the tampon tip again into the electrolyte, the tampon tip is moved back and forth or in small circles on the area to be rhodium-plated, applying very slight pressure. The article is rhodium-plated as long as the sufficiently soaked tampon tip is in contact with the article.

Metal depletion in the contact area is avoided by:

- the special composition of the electrolyte
- good movement of the tip on the part
- occasional renewed dipping of the tip into the electrolyte supply (approx. 2 - 3 times per minute)

Deformed and blunt tampon tips can be resharpened. Occasionally change the tip. A high consumption of tips results from a too high operating voltage because then the tips melt together due to overheating and the current passage is impeded.

Seal the tip with the cap prior to idle periods.

Posttreatment:

Rinse off the electrolyte rests in a recovery rinse (still rinse). Rinse the parts in deionized water and dry.

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#### **Note**

Our information relating to the storage stability refers to storage in closed original storage containers under the conditions stated on the label.

### **Precautionary Measures/Safety Hints**

For information on safety, please see the corresponding Material Safety Data Sheets! The valid accident prevention regulations and safety instructions must be observed.

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